

(3-Methoxyphenyl)acetic acid

A. R Choudhury and
T. N. Guru Row*

Solid State and Structural Chemistry Unit,
Indian Institute of Science, Bangalore 560 012,
Karnataka, India

Correspondence e-mail:
ssctng@sscu.iisc.ernet.in

Key indicators

Single-crystal X-ray study

$T = 293$ K

Mean $\sigma(\text{C}-\text{C}) = 0.002$ Å

R factor = 0.040

w R factor = 0.117

Data-to-parameter ratio = 11.5

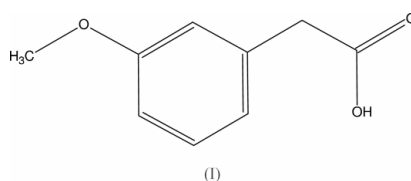
For details of how these key indicators were
automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $\text{C}_9\text{H}_{10}\text{O}_3$, also known as (*m*-methoxyphenyl)acetic acid, has been found to crystallize in the monoclinic space group $P2_1/c$ at room temperature. It is the starting material for the synthesis of a large number of 1,2,3,4-tetrahydroisoquinoline compounds [Nagarajan *et al.* (1985). *Indian J. Chem. Sect. B*, **24**, 83–97]. The compound forms dimers in the crystalline state, with $\text{O}-\text{H}\cdots\text{O}$ hydrogen bonds between the carboxyl groups, across a crystallographic centre of symmetry.

Received 5 July 2002

Accepted 9 July 2002

Online 19 July 2002



Experimental

The 98% pure compound was bought from Sigma Aldrich. Single crystals were grown from a mixture of ethyl acetate and hexane at room temperature, by slow evaporation. The compound crystallizes as prisms.

Crystal data

$\text{C}_9\text{H}_{10}\text{O}_3$

$M_r = 166.17$

Monoclinic, $P2_1/c$

$a = 16.266$ (3) Å

$b = 5.1024$ (11) Å

$c = 10.095$ (2) Å

$\beta = 90.73$ (1)°

$V = 837.8$ (3) Å³

$Z = 4$

$D_x = 1.317$ Mg m⁻³

Mo $K\alpha$ radiation

Cell parameters from 3478

reflections

$\theta = 2.4$ – 21.3 °

$\mu = 0.10$ mm⁻¹

$T = 293$ (2) K

Prism, colourless

$0.40 \times 0.35 \times 0.20$ mm

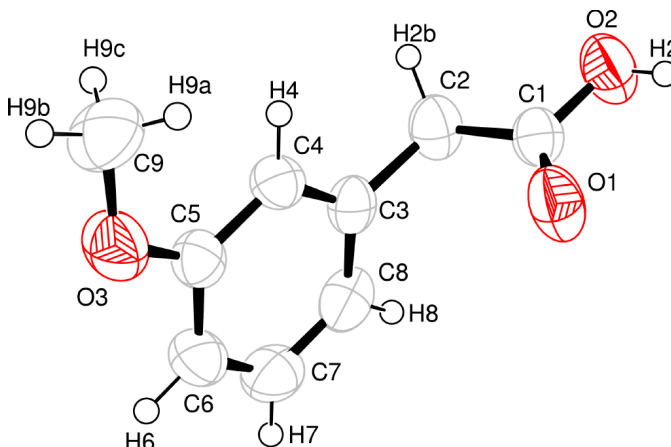


Figure 1

The molecular structure of the title compound, drawn with 50% probability displacement ellipsoids.

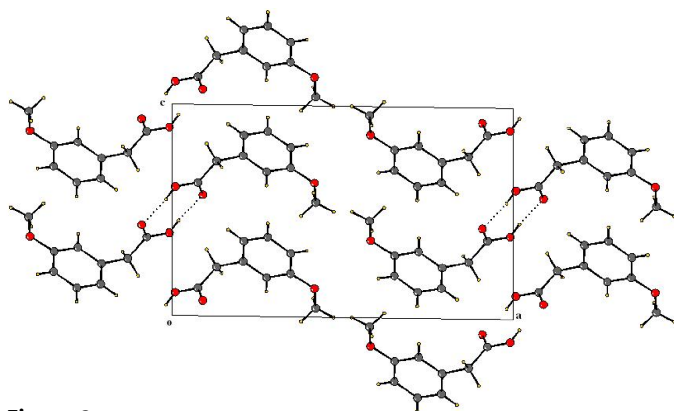


Figure 2
Packing diagram of the title compound, viewed down the *b* axis, showing O—H...O hydrogen-bonded dimers.

Data collection

Bruker SMART CCD area-detector diffractometer	1310 reflections with $I > 2\sigma(I)$
φ and ω scans	$R_{\text{int}} = 0.037$
Absorption correction: none	$\theta_{\text{max}} = 26.4^\circ$
6276 measured reflections	$h = -20 \rightarrow 20$
1714 independent reflections	$k = -6 \rightarrow 6$
	$l = -12 \rightarrow 12$

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.077P)^2 + 0.0041P]$
$R[F^2 > 2\sigma(F^2)] = 0.040$	where $P = (F_o^2 + 2F_c^2)/3$
$wR(F^2) = 0.117$	$(\Delta/\sigma)_{\text{max}} = 0.003$
$S = 1.03$	$\Delta\rho_{\text{max}} = 0.15 \text{ e } \text{\AA}^{-3}$
1714 reflections	$\Delta\rho_{\text{min}} = -0.16 \text{ e } \text{\AA}^{-3}$
149 parameters	
H-atom parameters constrained	

Table 1

Hydrogen-bonding geometry (\AA , $^\circ$).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O2-H2\cdots O1^i$	0.82	1.87	2.687 (2)	173

Symmetry code: (i) $-x, 1-y, 1-z$.

Data collection: *SMART* (Bruker, 1998); cell refinement: *SMART*; data reduction: *SAINT* (Bruker, 1998); program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *ORTEP-3* for Windows (Farrugia, 1997) and *CAMERON* (Watkin *et al.*, 1993); software used to prepare material for publication: *PLATON* (Spek, 1990).

The authors thank the Department of Science and Technology, India, for the data collection on the CCD facility set up under the IRFA-DST programme.

References

- Bruker (1998). *SMART* and *SAINT*. Bruker AXS Inc., Madison, Wisconsin, USA.
- Farrugia, L. J. (1997). *J. Appl. Cryst.* **30**, 565.
- Nagarajan, K., Talwalker, P. K., Kulkarni, C. L., Shah, R. K., Shenoy, S. J. & Pravu, S. S. (1985). *Indian J. Chem. Sect. B*, **24**, 83–97.
- Sheldrick, G. M. (1997). *SHELXL97* and *SHELXS97*. University of Göttingen, Germany.
- Spek, A. L. (1990). *Acta Cryst.* **A46**, C-34.
- Watkin, D. J., Pearce, L. & Prout, C. K. (1993). *CAMERON*. Chemical Crystallography Laboratory, University of Oxford, England.